

Friction Modifiers Optimization on Tribological Properties of Non-asbestos Organic (NAO) Brake Pad by DoE-Taguchi Method

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ABSTRACT

Friction modifiers play an important role in modifying the friction performances on the brake pad material. This study aims to determine the optimal friction modifier composition for obtaining a high friction coefficient (CoF) stability and low wear with the Taguchi method. Taguchi's L_8 orthogonal array consists of five factors, and two levels were used to design and perform the experiment. The tribological properties (CoF stability and wear) were tested using the JF160 Chase friction tester according to the SAE J661 standard. According to the S/N ratio and ANOVA analysis, the controllable factor of h-BN has a significant effect on CoF stability. While graphite and h-BN have a dominant influence on wear. Contribution of 4 vol.% h-BN, 4 vol.% MoS₂, 4 vol.% Al₂O₃, 2 vol.% graphite, and 3 vol.% SiO₂ is an optimal composition to obtain high CoF stability. Meanwhile, the contribution of graphite (8 vol.%), h-BN (2 vol.%), MoS₂ (4 vol.%), Al₂O₃ (2 vol.%), and SiO₂ (3 vol.%) can be an optimal composition on lower wear. The predictive value of CoF stability and wear on the optimal level, as made by the Taguchi method, is consistent with the confirmation test within a 95% confidence interval (CI).

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1. INTRODUCTION

The brake pad materials usually comprise more than 10-20 different ingredients [1-2], which are grouped as binders, reinforcements, fillers, and friction modifiers [3-7]. Friction modifiers are one of the brake pad ingredients that play an important role in modifying friction and wear performances as they serve both as abrasives and solid lubricants.

Abrasives increase friction and wear rate. Abrasives usually are added around 1-8 vol.% to the brake pad [9]. High content of abrasives in the brake pad materials causes instability of friction coefficient, damage to the disc, and noise during the braking [10]. On the contrary, solid lubricants are beneficial to reduce abrasion between pad and disc, improve wear resistance, dampen vibration, and decrease noise [8]. Hence, a careful selection of the

composition between abrasives and solid lubricants is crucial since they often have the opposite effect on the friction performances generated during the braking.

There are commonly used materials as abrasives and solid lubricants in the brake pad. The abrasive materials are silicon carbide (SiC), zirconium oxide (ZrO₂), zirconium silicate (ZrSiO₄), aluminum oxide (Al₂O₃), and silicon dioxide (SiO₂) [11]. Meanwhile, antimony sulfide (Sb₂S₃), graphite, copper sulfide (CuS), hexagonal boron nitride (h-BN), and molybdenum disulfide (MoS₂) are frequently used as solid lubricants in the brake pad [12].

In previous studies, materials such as ZrSiO₄ [13], Al₂O₃ [14-16], SiO₂ [17], SiC [13], h-BN [18,19], Sb₂S₃ [20], MoS₂ [21], and graphite [22] have been studied. These studies only focused on understanding the individual effects of abrasives and solid lubricants on the brake friction performances experimentally. However, no comprehensive study ever discussed statistically and experimentally between abrasives and solid lubricants (friction modifiers) on brake friction performances with the Taguchi method.

Selection of the proper friction modifier composition is necessary because it directly affects brake friction performances. The availability of a significant number of the friction modifiers compositions has a complex relationship with the braking results. Consequently, composition selection is a difficult task and requires a great number of experiments. Taguchi method can be used to reduce the number of experiments efficiently [23-25]. This study aims to determine the

optimal friction modifier composition (namely: Al₂O₃, SiO₂, MoS₂, h-BN, and graphite) to obtain high CoF stability and low wear using Taguchi Method.

2. EXPERIMENTAL

2.1 Design of experiment (DOE)

Taguchi's design was applied to design an optimum level of friction modifiers for maintaining the variance at high CoF stability and low wear. In the present study, five controllable factors of friction modifier materials (graphite, hexagonal boron nitride (h-BN), molybdenum disulfide (MoS₂), aluminum oxide (Al₂O₃), and silicon dioxide (SiO₂)) and two levels for each of these factors were selected with Taguchi design, as shown in Table 1. According to the optimal value reported in the article published on the brake friction performance, the levels of each factor were selected. Taguchi L₈ (2³) orthogonal array with eight number of experiments was selected to experiment. Furthermore, the experimental results were analyzed using the statistical software MINITAB 17. The S/N ratio is determined from the experimentally observed value and then statistically analyzed by ANOVA (Analysis of Variance). Table 2 shows the experimental layout of the Taguchi L₈ orthogonal arrays.

Table 1. Experimental factors and their levels.

Factor	Level (vol.%)	
	1	2
Graphite	4	8
h-BN	2	4
MoS ₂	2	4
Al ₂ O ₃	2	4
SiO ₂	3	5

Table 2. Experimental layout of Taguchi L₈ (2³) orthogonal array and results for two levels of parameter.

Exp. No	Control Factor (vol.%)					Output Results			
	Graphite	h-BN	MoS ₂	Al ₂ O ₃	SiO ₂	CoF stability	S/N ratio (dB) for CoF stability	Wear (%)	S/N ratio (dB) for wear
1	4	2	2	2	3	79.06	37.95	10.05	-20.07
2	4	2	2	4	5	77.46	37.78	8.02	-18.08
3	4	4	4	2	3	84.12	38.49	9.70	-19.74
4	4	4	4	4	5	89.30	39.02	15.91	-24.03
5	8	2	4	2	5	77.74	37.81	5.84	-15.27
6	8	2	4	4	3	81.56	38.23	5.52	-14.81
7	8	4	2	2	5	84.04	38.49	9.90	-19.91
8	8	4	2	4	3	83.99	38.48	8.73	-18.79

The brake pad sample was also comprised of binder (i.e., phenolic resin), fillers (i.e., friction dust and BaSO₄), reinforcements (i.e., rockwool, cellulose, and PAN fibers), and copper. These ingredients were kept constant (78% vol.%), and MgO was varied in volume fraction to balance the total composition of the sample. The Taguchi design was used with the assumption that there was no interaction effect among other ingredients on tribological properties, and the composition MgO did not affect tribological properties.

2.2 Sample preparation and tribological test

The samples were manufactured with a five-step process: weighing, mixing, hot pressing, post-curing, and finishing. The manufacturing details of the sample were explained in our previous study [26-27]. Tribological testing of the sample was carried out using the JF160 chase friction tester by following SAE J661 standard procedures. The test sample with a dimension of 25.4 x 25.4 x 7 mm (Fig. 1) was positioned on 140 mm an inner radius of friction drum. The sample was then pressed hydraulically against the friction drum under a normal load of 660 N. The rotating speed of the friction drum was 411 rpm, and the linear sliding velocity of the sample was 6.028 m/sec. The schematic of the JF160 chase friction tester is given in Fig. 2. Each sample has followed seven stages of the testing procedure, as shown in Table 3. The tribological test was repeated three times for each sample parameter. Samples were then weighed, and their thickness was measured before and after the test. After test procedures were completed, CoF

stability and wear were then calculated and taken as a response for performance optimization. CoF stability and wear were calculated using the following equations:

$$CoF\ stability = \frac{\mu_p}{\mu_{max}} \quad (1)$$

$$W = \frac{w_o - w_1}{w_o} \times 100\% \quad (2)$$

Where μ_{max} is a maximum friction coefficient, μ_p is the average friction coefficient, w_o and w_1 are the average weight loss of the sample before and after the friction test.

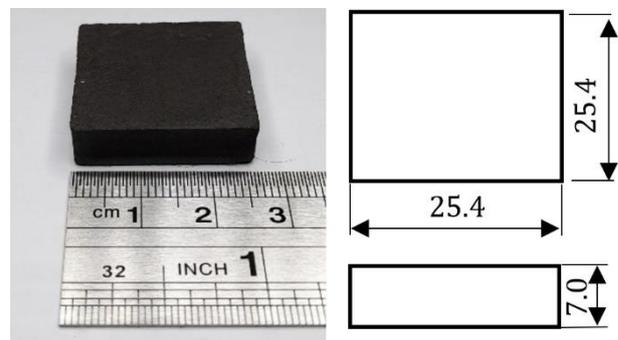


Fig. 1. Friction test sample.

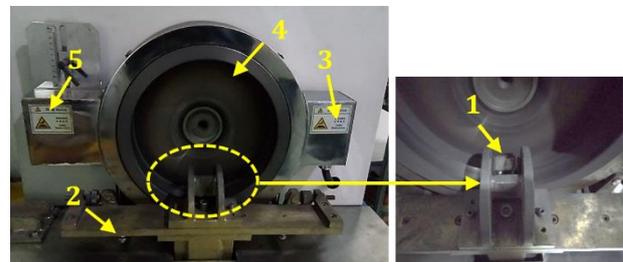


Fig. 2. JF160 chase friction tester: (1) sample, (2) loading arm, (3) blower, (4) friction drum, and (5) heater.

Table 3. Experimental procedure of SAE J661.

Cycles	Load (N)	Speed (rpm)	Temperature (°C)	Braking Mode
Burnish	440	312	< 95	Continues braking for 20 min
Initial baseline	660	411	82 - 104	Intermittent braking 10 sec On, 20 sec Off for 20 applications
Fade-I	660	411	93 - 261	Continues and heater On
Recovery-I	660	411	260 - 92	Continues and blower On
Wear	660	411	193 - 204	Intermittent braking 10 sec On, 20 sec Off for 100 applications
Fade-II	660	411	93 - 345	Continues and heater On
Recovery-II	660	411	316 - 92	Continues and blower On
Final baseline	660	411	104 - 82	Intermittent braking 10 sec On, 20 sec Off for 20 applications

3. RESULTS AND DISCUSSION

3.1 The S/N ratio and ANOVA for CoF stability

Based on the Taguchi method, response variation using the S/N ratio is crucial because it can minimize variations in quality characteristics due to uncontrollable parameters. CoF stability level must be as high as possible in the brake pad materials as it contributes to smooth braking and avoids jerks during braking applied [27]. Therefore, CoF stability response was considered a quality characteristic using the S/N ratio “the large the better”. The S/N ratio of the quality characteristics “the large the better” is calculated by equation (3) [23].

$$\frac{S}{N} \text{ ratio} = -10 \log \left(\frac{1}{n} \sum_{i=1}^n \frac{1}{y_i^2} \right) \quad (3)$$

Where y_i = experimental results, and n = number of experiments. The average values of CoF stability measured from the experiment test and their S/N ratio are presented in Table 2. Meanwhile, Table 4 shows the S/N ratio of CoF stability for friction modifiers. According to the rank order, h-BN is the most influential, followed by MoS₂ and Al₂O₃. In contrast, graphite and SiO₂ are affected CoF stability the least.

Table 4. Responsible table for S/N ratio of CoF stability of friction modifiers (the large the better).

Level	Graphite	h-BN	MoS ₂	Al ₂ O ₃	SiO ₂
1	38.31	37.95	38.18	38.19	38.29
2	38,25	38.62	38.39	38.38	38.27
Delta	0.06	0.68	0.21	0.19	0.02
Rank	4	1	2	3	5

Fig. 3 reveals the main effect plot for the S/N ratio of CoF stability. In the main effect plot, if the line for a certain factor is near horizontal, the factor has no significant effect on the response. On the contrary, the factor with the highest slope of the line has the most considerable effect. As presented in Fig. 3, the optimal CoF stability for the vol.% contribution is obtained on 4 vol.% graphite, 4 vol.% h-BN, 4 vol.% MoS₂, 4 vol.% Al₂O₃, and 3 vol.% SiO₂. The controllable factor of h-BN significantly affects on CoF stability because h-BN is a good lubricant at elevated temperature, improving heat resistance, and lubrication ability on the friction material [29]. Yanar *et al.* [30]

reported that the h-BN composition of 2.5 to 3 vol.% in non-asbestos brake pad composites resulted in a more stable friction coefficient. A similar result has been observed in the previous study [31] that 2 vol.% h-BN with graphite in Cu-based friction composite can stabilize friction coefficient and wear resistance. h-BN could contribute to attenuating the effect of varying temperatures on frictional behavior.

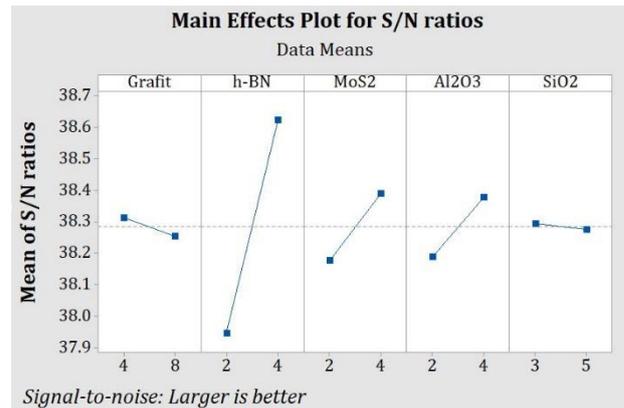


Fig. 3. Main effects plot for S/N ratios of CoF stability of friction modifiers.

MoS₂ and Al₂O₃ have more effect on the CoF stability compared to graphite and SiO₂. MoS₂ has self-lubricating capabilities, and it can prevent the wear debris from sticking to the disc surface. MoS₂ prevents the formation of heterogeneous transfer films, leading to instability in the friction coefficient [21,32]. On the other hand, the addition of Al₂O₃ in the friction material causes the sample to be less sensitive to temperature increases during sliding. Thus, it resulted in a stable friction coefficient [33]. Singh *et al.* [34] reported that the presence of 5 wt.% Al₂O₃ in non-asbestos brake composite resulted in the highest friction stability (0.87) and friction variability (0.48). It is due to Al₂O₃ as an abrasive, which can remove pyrolysis (unwanted coating) and renews the damaged (degraded) friction surface. Therefore, it can increase the integrity of friction film and friction stability [16].

Graphite and SiO₂ at low factor levels (4 vol.% graphite and 3 vol.% SiO₂) have an insignificant effect on the CoF stability. The low content of graphite in the friction material causes the occurrence of adhesion. It prevents the formation of the friction layer since only a few graphites may be distributed on the friction surface [35]. Thereby, the low content of graphite increases the roughness of the surface

and instability of the friction coefficient. Meanwhile, SiO₂ has an insignificant influence on the optimum friction coefficient due to its abrasive nature that causes pits formation on the wear surface. Pits formation on the wear surface increases the rough surface, friction fluctuation, and higher wear rate [34].

Table 5 shows the ANOVA analysis results for the S/N ratio of CoF stability. The ANOVA analysis is evaluated for a confidence level of 95% (α = 0.05). A P-value less than 0.05 is considered to have a statistically significant contribution to the response parameter. In contrast, a P-value of more than 0.05 shows an insignificant contribution to response. The percentage contribution of P is calculated by equation (4).

$$\% \text{ contribution} = \frac{\text{Seq SS}_{\text{factor}}}{\text{Seq SS}_{\text{total}}} \times 100\% \quad (4)$$

Table 5. ANOVA for S/N ratio of CoF stability (DF: Degree of freedom, Seq SS: Sum of squares, Adj MS: Adjusted mean of squares, F: ratio, P: value, Cont: Contribution).

Source	DF	Seq SS	Adj MS	F	P	Cont (%)
Graphite	1	0.007	0,007	0.09	0.798	0.57
h-BN	1	0.913	0.912	10.98	0.048	73.16
MoS ₂	1	0.089	0.089	1.070	0.409	7.14
Al ₂ O ₃	1	0.072	0.071	0.860	0.451	5.76
SiO ₂	1	0.000	0.000	0.01	0.935	0.06
Error	2	0.166	0.083			13.32
Total	7	1.247				100

As shown in Table 5, the controllable factor of h-BN is found to be significant (p-value < 0.05) and other factors insignificant (p-value > 0.05) for the CoF stability at 95% confidence level. The contribution percentage of each factor is h-BN (73,16%), MoS₂ (7,14), Al₂O₃ (5.76), graphite (0.57), and SiO₂ (0.06%).

3.2 S/N ratio and ANOVA for wear

The brake pad wear should be as low as possible because it increases the brake pad’s service life. The concept of “the lower the better” was used to calculate the S/N ratio of wear quality characteristics. The S/N ratio of wear is calculated by equation (5).

$$\frac{S}{N} \text{ ratio} = -10 \log \left(\frac{1}{n} \sum_{i=1}^n y_i^2 \right) \quad (5)$$

Where y_i = experimental results, and n = number of experiments. Table 6 shows the S/N ratio of wear for friction modifiers. As shown in Table 6, h-BN has the most effect on low wear, followed by graphite, SiO₂, and MoS₂. Meanwhile, Al₂O₃ has a much lower effect on low wear.

Table 6. Responsible table for S/N ratio of wear of friction modifiers (the lower the better).

Level	Graphite	h-BN	MoS ₂	Al ₂ O ₃	SiO ₂
1	-20.48	-17,06	19,22	-18,75	-18,35
2	-17,20	-20,62	18,46	-18,93	-19,32
Delta	3,28	3,56	0,76	0,18	0,97
Rank	2	1	4	5	3

Fig. 4 represents the main effect plot for the S/N ratio on the wear of friction modifiers. Wear decreases with an increasing composition of graphite and MoS₂ and decreasing composition of h-BN, SiO₂, and Al₂O₃. Similar results were observed in the previous study [31]. The author reported that the content of 2 vol.% h-BN and 8 vol.% graphite in Cu-based composite resulted in friction stability and wear rate.

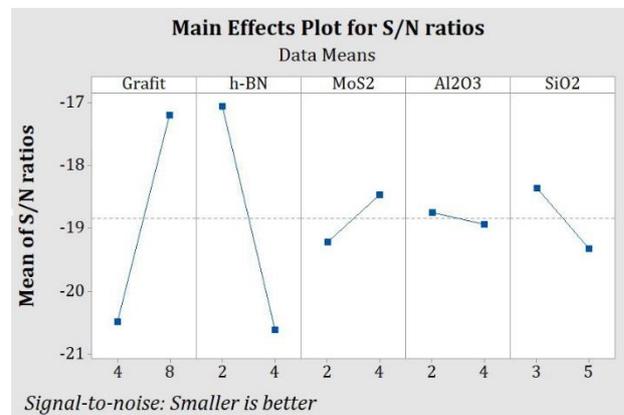


Fig. 4. Main effects plot for S/N ratios of wear of friction modifiers.

The h-BN has the most influence on minimum wear. This finding is attributed to the fact that h-BN can be used as a lubricant to optimizes friction reduction over a wide range of temperatures [19]. Graphite also shows a dominant influence on low wear after h-BN. The increase of graphite content on the friction material causes more graphite to be distributed on the friction surface. This condition can prevent adhesion and contribute to forming a

friction layer on the friction interface [36-37]. The friction layers prevent the hard particles from sticking in the friction interface [35] and inhibit the micro-welding of metallic ingredients [38], thereby protecting the surface from excessive wear [12,39]. In the previous studies, the increase in graphite content (0 - 10 vol.%) in friction composite significantly reduces wear rates [31]. Yang *et al.* [35] also reported that graphite enhancement (2.5 -10 wt.%) could reduce the average friction coefficient, friction fluctuation, and wear rate.

The minimum wear for the vol.% contribution is obtained on h-BN (2 vol.%), graphite (8 vol.%), SiO₂ (3 vol.%), MoS₂ (4 vol.%), and Al₂O₃ (2 vol. %).

Table 7. ANOVA for S/N ratio of wear (DF: Degree of freedom, Seq SS: Sum of squares, Adj MS: Adjusted mean of squares, F: ratio, P: value, Cont: Contribution).

Source	D F	Seq SS	Adj MS	F	P	Cont (%)
Graphite	1	21.574	21.574	4.32	0.173	35.96
h-BN	1	25,344	25.344	5.08	0.153	42.24
MoS ₂	1	1.141	1.141	0.23	0.680	1.90
Al ₂ O ₃	1	0.066	0.065	0.01	0.919	0.11
SiO ₂	1	1.885	1.884	0.38	0.602	3.14
Error	2	9.985	4.992			16.64
Total	7	59.994				

Table 7 exhibits the ANOVA analysis results for the S/N ratio of wear. It is clearly seen that all of the controllable factors have an insignificant effect at a 95% confidence level (p-value > 0.05) on the minimum wear of friction modifiers. The contribution percentage of h-BN (42.24%) is more influential than graphite (35.96%). The contribution percentages of MoS₂, Al₂O₃, and SiO₂ on wear are 1.90%, 0.11%, and 3.14%, respectively.

3.3 Confirmation test within confidence interval (95% CI)

The confirmation test is a crucial step to verify the results from the Taguchi design. In this study, the confirmation test was utilized on the optimal level of all the controllable factors. Three samples were then prepared under the optimal levels of CoF stability and wear. The optimum level of CoF stability is performed on graphite (4 vol.%), h-BN (4 vol.%), MoS₂ (4 vol.%), Al₂O₃ (4 vol.%), and SiO₂ (3 vol.%). Meanwhile, the optimal level of wear is

carried out on the composition of 8 vol.% graphite, 2 vol.% h-BN, 4 vol.% MoS₂, 2 vol.% Al₂O₃, and 3 vol.% SiO₂. Table 8 represents the confirmation test results on the optimal level.

Table 8. Confirmation test results on the optimal level.

Variable	Sample			Average
	1	2	3	
CoF stability	88.431	89.25	91.552	89.738
Wear (%)	5.276	4.767	6.038	5.36

The predictive value of the optimal level can be calculated by the Taguchi method, and this value is consistent with the confirmation test. The confidence interval (CI) is used for the confirmation test [23] and calculated by equation (6).

$$CI = \pm \sqrt{\frac{F(1, n_2) \times MSe}{N_e}} \quad (6)$$

Where F(1, n₂) is F-table value on confidence level desired, MSe is the sum of the mean squared error calculated in the ANOVA table of S/N ratio/mean. N_e is an effective number of replications. N_e is calculated by equation (7).

$$N_e = \frac{\text{Total number of experiment}}{1 + \text{DoF all factors in estimate of mean}} \quad (7)$$

Table 9 represents the predictive value results (95% CI) for the S/N ratio on the optimal level of CoF stability and wear. From these results, the following equation model can be drawn:

Table 9. Predictive value (95% CI) for S/N ratio on the optimal level.

Variable	Predicted value (S/N ratio)	F(1,n2)	MSe	Ne	95% CI
CoF stability	38.854	13.01	0.166	2.667	± 0.899
Wear	19.176	10.02	9.985	2.667	± 6.125

Predicted value - CI ≤ predicted value ≤ predicted value + CI.

Therefore, the equation model of the S/N ratio prediction on the optimal level is

$$37.955 \leq 38.854 \leq 39.753 \text{ for CoF stability, and } 13.051 \leq 19.176 \leq 25.301 \text{ for wear}$$

Table 10. Predictive value (95% CI) for CoF stability and wear on the optimal level.

Variable	Predicted value (mean)	F(1,n2)	MSe	Ne	95% CI
CoF stability	87.660	12.79	7.654	2.667	± 6.058
Wear	6.025	6.55	8.578	2.667	± 4.589

Table 10 shows the predictive value results (95% CI) for CoF stability and wear on the optimal level. Consequently, the equation model below may be formulated:

Predicted value - CI ≤ predicted value ≤ predicted value + CI.

The equation model for mean prediction value on the optimal level is

$81.602 \leq 87.660 \leq 93.718$ for CoF stability, and $1.436 \leq 6.025 \leq 10.614$ for wear

Based on the above results, the Taguchi method's predictive value is consistent with the confirmation test. On the optimal level, the predictive value and confirmation test results of the CoF stability and wear are presented in Figs. 5 and 6.

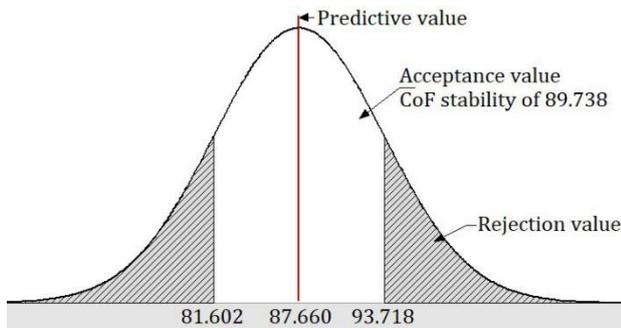


Fig. 5. Mean interval of CoF stability within 95% CI.

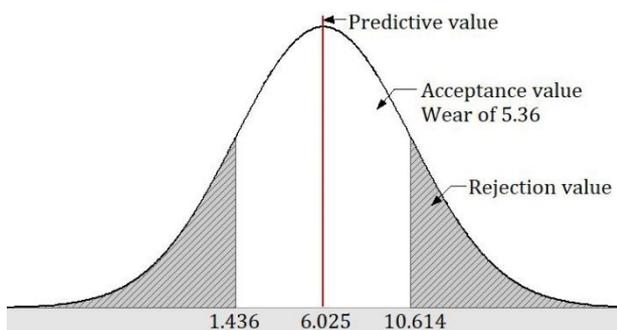
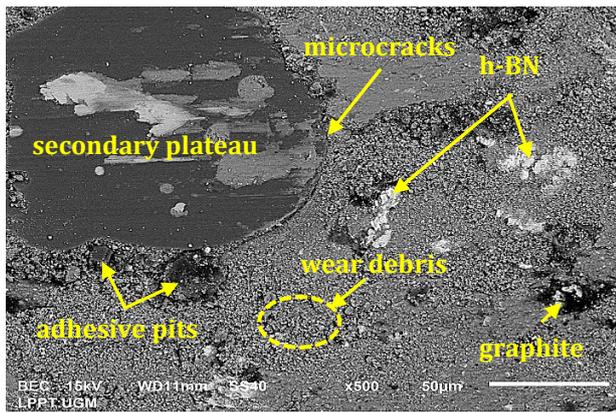


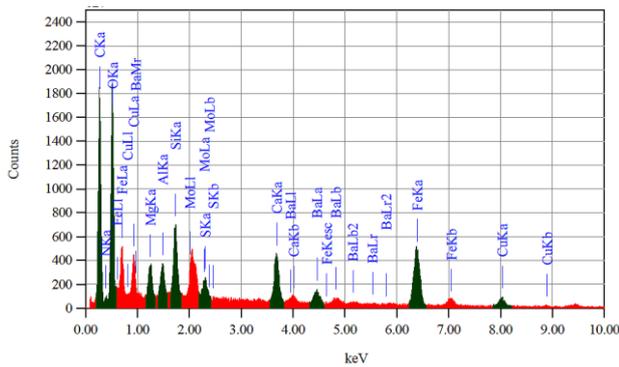
Fig. 6. Mean interval of wear within 95% CI.

3.4 Morphology of worn surface

SEM image reveals the different characteristics of secondary/primary contact plateau, adhesive pits, wear debris, and microcracks related to the wear mechanism during the tribological test. SEM images and EDS analysis of the sample for CoF stability and wear on the optimal level are presented in Figs. 7 and 8. Figure 7a exhibits the worn surface of the sample for the optimal level of CoF stability. It is seen that the worn surface is relatively rougher. Besides, the secondary plateau, adhesive pits, microcracks, wear debris, and the agglomeration of graphite/hBN are also observed. The secondary plateau is formed by layers adhered [40] and finer debris compacted to the surface under shear force, normal force, and heat friction, simultaneously [41]. Meanwhile, the formation of microcracks is a result of thermal shock stress [42]. The sample surface is subjected to force and friction heat, which generated an unstable temperature and force. Thus, there is a difference in expansion heat in some areas of the friction interface, resulting in microcracks on the worn surface (Fig. 7a). Microcracks is a type of fatigue wear [4]. Meanwhile, adhesive pits are formed on the worn surface due to shearing force, indicating adhesive wear has occurred. Therefore, the sample wear mechanism is abrasive wear, adhesive wear, and fatigue wear. From the EDS analysis as shown in Fig. 7b, elements were detected on the worn surface, namely: carbon (C), nitrogen (N), oxygen (O), magnesium (Mg), aluminum (Al), silicon (Si), iron (Fe), molybdenum (Mo), calcium (Ca), and copper (Cu). These elements are originated from the sample ingredients. The higher concentration of elements on the worn surface is found on the elements of C (30.83%), O (28.93%), N (9.32%), and Fe (13.83%). C and O elements at a higher level indicate oxidation on the sample, which might occur from the thermal oxidation of organic and inorganic ingredients as graphite and MoS₂. In previous studies, graphite decomposed into the gas at elevated temperature [35]. Whereas, MoS₂ reacted with oxygen and transformed into oxide form (MoO₃) at 540 °C [22,43]. Fe element detected on the worn surface is possibly transferred from friction drum (cast iron) due to thermal-oxidation during friction test. The thermal oxidation on the disc surface causes film transfer (containing Fe₂O₃) to the sample surface. This finding is in line with previous studies [27,44,45].



(a)

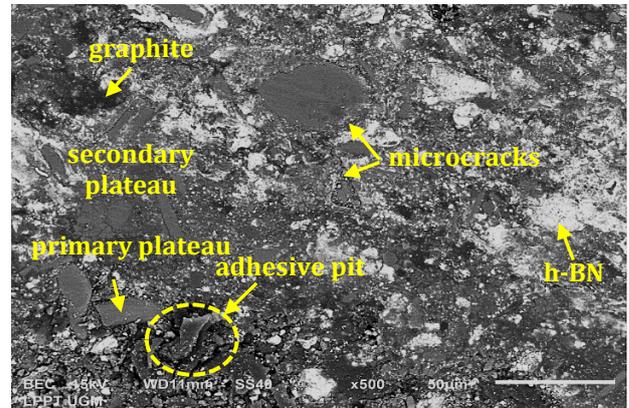


(b)

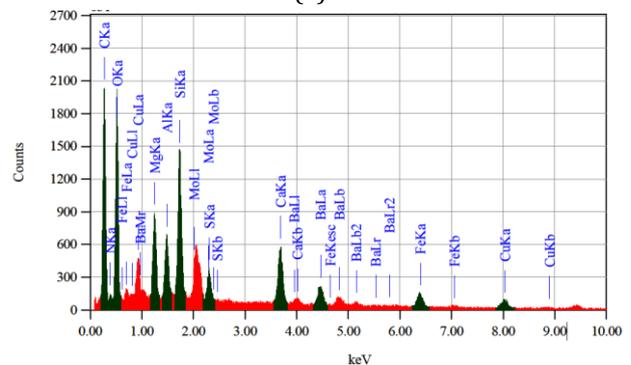
Fig. 7. Sample on the optimal level of CoF stability (a) worn surface morphology, and (b) EDS analysis.

Figure 8a shows the worn surface sample for the optimal level of wear. The worn surface is relatively smooth and visibly covered by a secondary plateau. Meanwhile, microcracks, wear debris, adhesive pits, primary plateau, graphite, and h-BN are also revealed on the worn surface. Primary plateau is formed from fiber trapped in the friction interface under shearing force. Figure 8b represents the EDS analysis, the area of worn surface is composed of some elements, including carbon (C), oxygen (O), nitrogen (N), iron (Fe), molybdenum (Mo), silicon (Si), aluminum (Al), magnesium (Mg), calcium (Ca), and copper (Cu). The higher concentration is found in the element of C (32.14%), O (28.95%), and N (11.97%), while lower concentration is obtained on the element of Al (1.74%), Si (4.29%), Fe (3.13%), and Mo (1.07%). These results indicate that the worn surface was dominated by compacted wear debris containing C, O, and N elements. These elements originated from the ingredients sample and oxidation, i.e., graphite, h-BN, MoS₂, Al₂O₃, and SiO₂. The wear debris containing many C-elements promotes a more extensive secondary plateau, which is responsible for improving wear resistance on the

sample. Therefore, the mechanism wear of this sample is abrasive wear, adhesive wear, and fatigue wear.



(a)



(b)

Fig. 8. Sample on the optimal level of wear (a) worn surface morphology, and (b) EDS analysis.

4. CONCLUSIONS

This experiment study determines the optimal level of friction modifiers on the tribological properties of non-asbestos organic brake pad using the Taguchi method. The results are summarized as follows:

1. The controllable factor of h-BN has a significant effect on the high CoF stability; while graphite, MoS₂, Al₂O₃, and SiO₂ are insignificant. Contribution of 4 vol.% h-BN, 4 vol.% graphite, 4 vol.% MoS₂, 4 vol.% Al₂O₃, and 3 vol.% SiO₂ can be an optimal composition of friction modifiers in non-asbestos organic (NAO) brake pad material to obtain a high CoF stability.
2. The controllable factor of h-BN and graphite have the most influence on lower wear. On the other hand, the effect of MoS₂, Al₂O₃, and SiO₂ is lower. Contribution of 8 vol.% graphite, 2 vol.% h-BN, 4 vol.% MoS₂, 2 vol.% Al₂O₃, and 3 vol.% SiO₂ is an optimal level of friction modifiers to obtain a low wear.

3. The CoF stability and wear predicted value by utilizing the optimal level of the Taguchi optimization method is consistent with the confirmation test (CoF stability mean value of 89.738 and wear is 5.36%) within the 95% confidence interval (CI).

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